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1. Teach students the predominant separation processes used in chemical engineering [1-8] 2. Introduce students to chemical engineering processes and equipment [1-7] 3. Show students how previous work in mathematics and physics is useful to them [2-6, 8] 4. Show students how the design of separation units impacts the environment [6] 5.

**TEXTBOOKS/REQUIRED MATERIAL: Separation Process Principles**

Separation Processes • Absorption • Solutes removed from a gas into a liquid • Solutes removed from liquid into gas is called stripping or desorption • Distillation • Thermal vapor-liquid separation processes (Ch 11); vapor phase generated from liquid • Liquid-liquid extraction • Solute extracted from liquid A into an immiscible liquid B (a solvent)

**Seader & Henley, Separation Process Principles**

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Dedicated to helping students and faculty use more active learning methods in their engineering courses.

**Textbook: Separation Process Principles (3rd Edition)**

Solution manual An Introduction to Mass and Heat Transfer : Principles of Solution manual Separation Process Principles (3rd Ed, Seader, Henley & Roper) Separation Process Principles 3rd ED ( Instructor's Solutions Manual ) Authors, Seader, Henley The Instructor Solutions manual is available in PDF format.

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Seader and Henley stress the viewpoint of unifying the rate-based approach and the equilibrium-based approach in a course that systematically proceeds through the separation operations after initial chapters on the fundamentals of diffusion and mass transfer (Ch.3) and on physical equilibrium (Ch. 2).

**Separation process principles - J. D. Seader, Ernest J**

From advances in the practice of separation operations in chemical engineering to corresponding changes in the curriculum, much has happened in the seven years since the publication of Seader and Henley's first edition of Separation Process Principles, including:

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1 1. Study Questions To gain the best understanding of the material presented in, "Separation Process Principles", 2nd Edition, by J.D. Seader and Ernest J. Henley, print these Study Questions and write in your answers as you read.

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Exercise 1.3 Subject: Separation of a mixture requires a transfer of energy to it or the degradation of its energy. Given: The first and second laws of thermodynamics. Find: Explain why the separation of a mixture requires energy. Analysis: As an example, consider the isothermal minimum (reversible) work of separation of an ideal binary gas mixture.

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